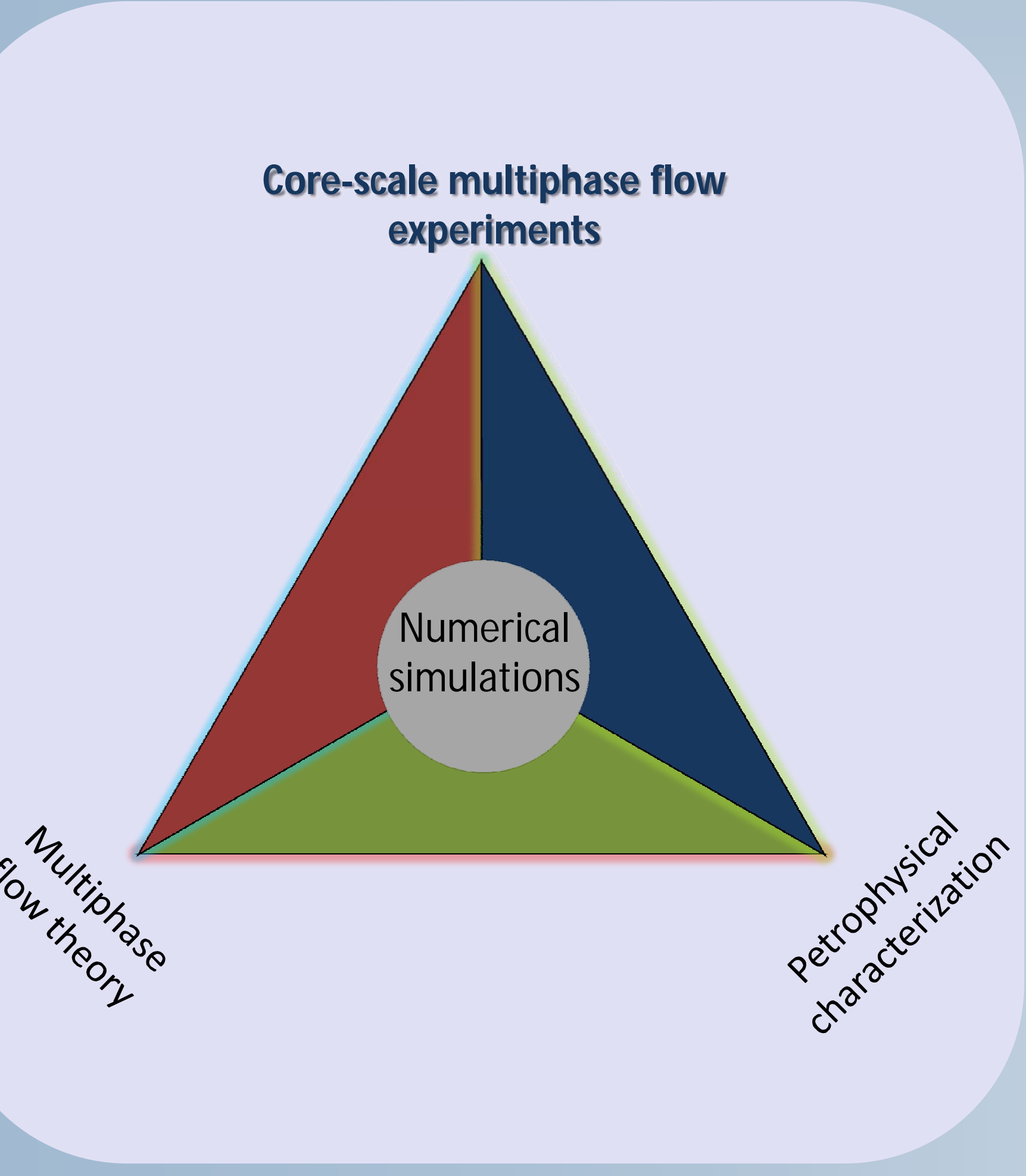
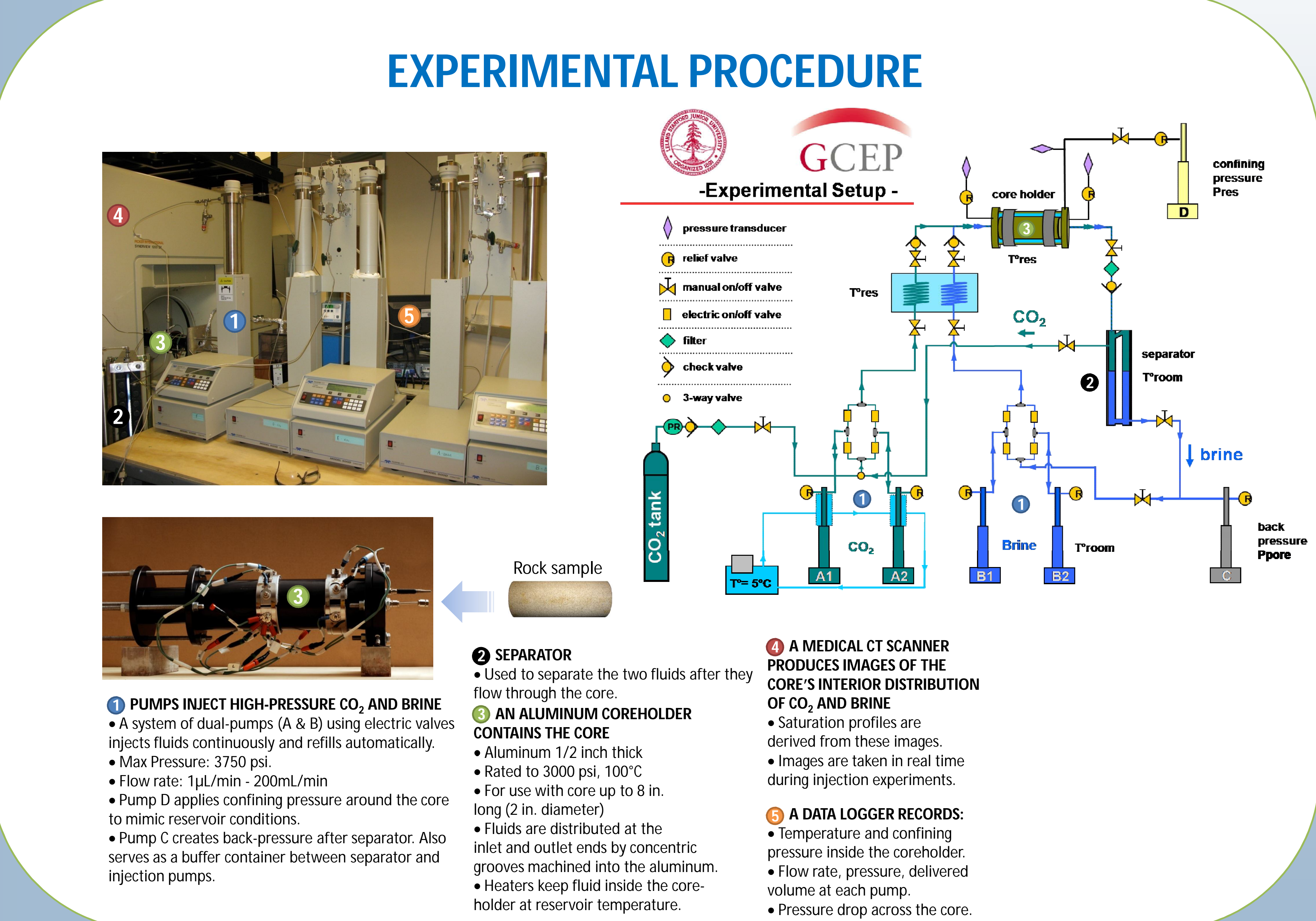
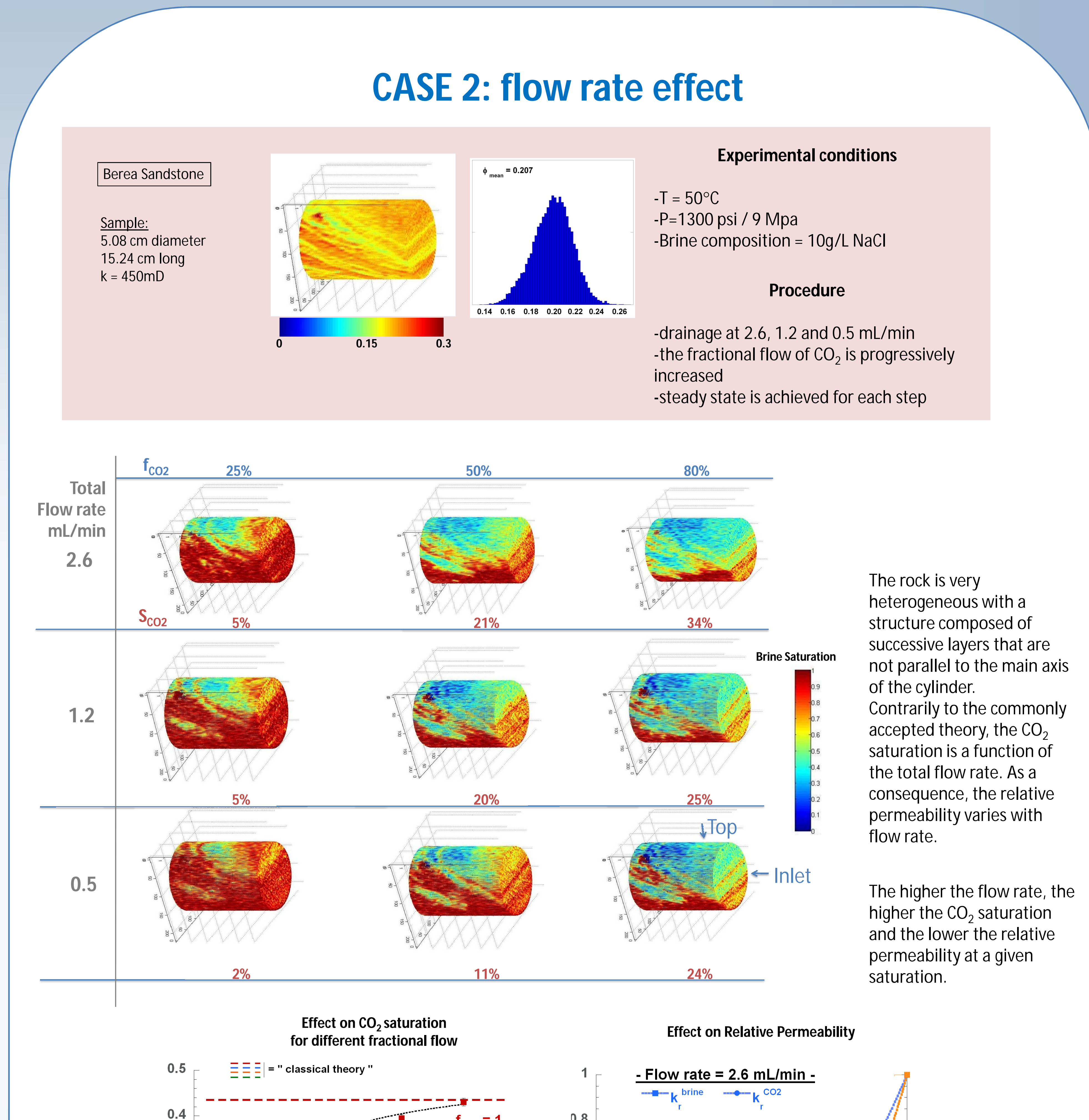
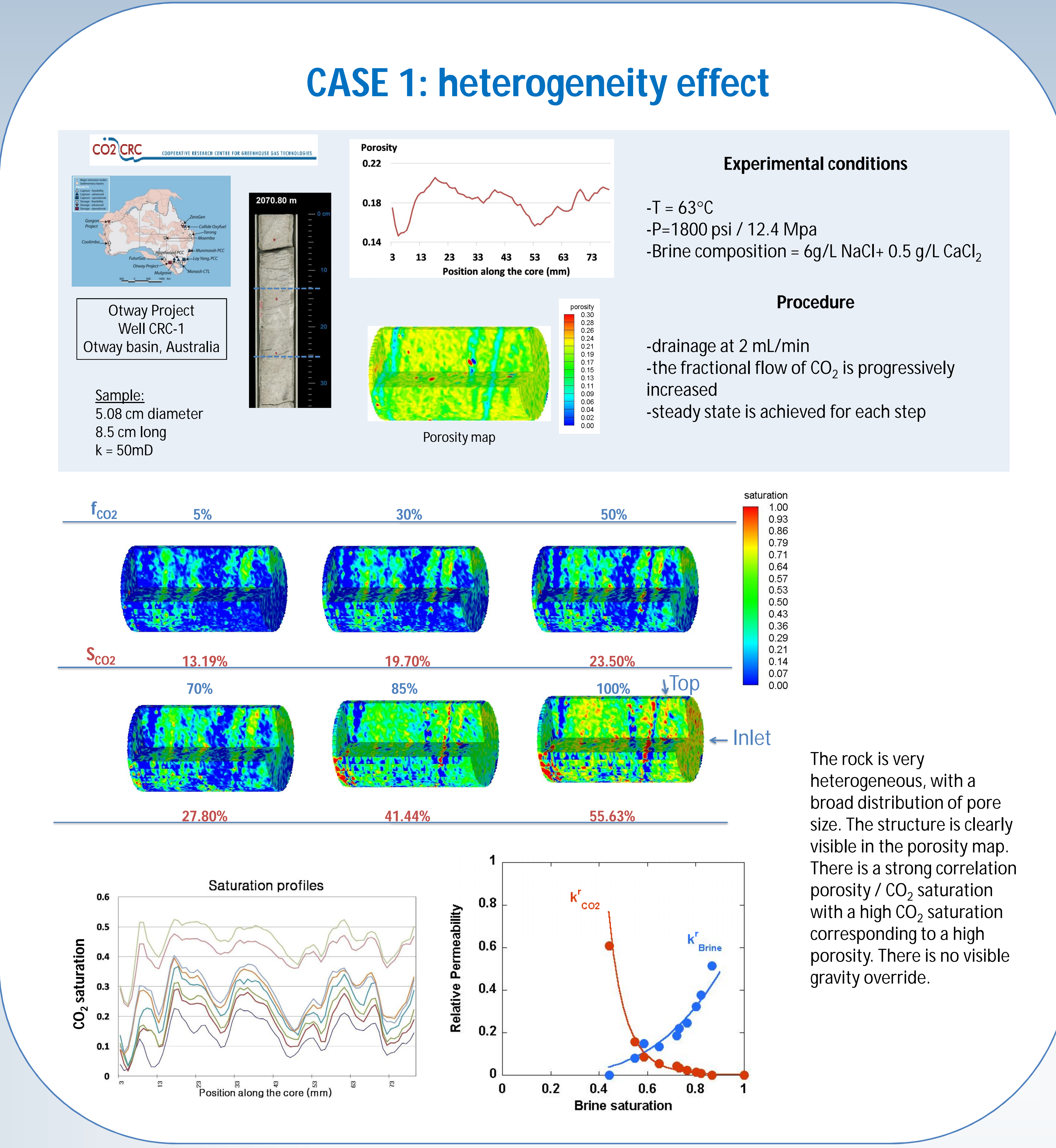


**THE SEQUESTRATION LAB**  
The goal of the Sequestration Lab is to develop the ability to predict the spatial and temporal distribution of CO<sub>2</sub> saturation and trapping through an improved understanding of the pore and core scale physics over the life cycle of a sequestration project.

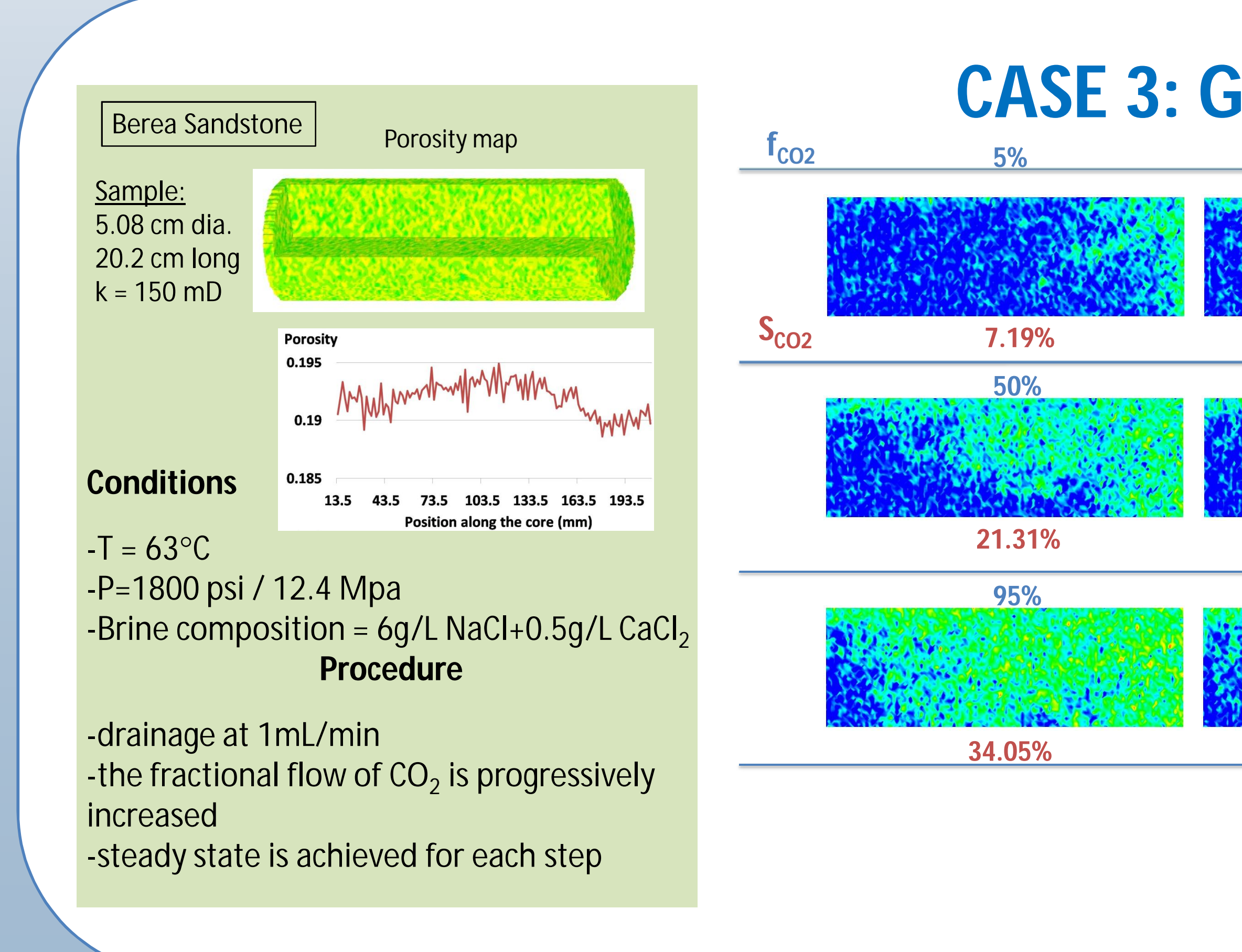


**INTRODUCTION / BACKGROUND**  
*Relative permeability* is a key concept for carbon dioxide storage. Defined in multi-phase flow in porous media as the ratio of effective permeability of a particular fluid at a particular saturation to absolute permeability of that fluid at total saturation, relative permeability controls the majority of the phenomena that are met when two (or more) fluids are flowing through rocks. Studying the relative permeability properties of the CO<sub>2</sub>-brine system in deep saline aquifers is fundamental to answer important questions that are met *in the field*. For instance:  
- What should be the pressure in the injection zone?  
- How big the plume is going to be?  
- How fast would CO<sub>2</sub> leak up a fault under buoyancy?  
- How to maximize sweep efficiency (storage capacity)?  
  
*In the laboratory*, relative permeability experiments associated with fine rock characterizations are performed in order to address more fundamental issues. For instance:  
- What are the trapping mechanisms?  
- How do the external factors influence relative permeability (pressure, temperature, injection rates, rock properties/structure)?



Thermophysical properties of brine and CO<sub>2</sub> at experimental conditions

| Temperature                    | 5°C  | Room T (21°C)  | 50°C  | 63°C  |
|--------------------------------|--|--|---|---|
| <b>Brine @1700 psi</b>         | Liquid<br>μ = 1.50 cP<br>d = 1.004 g/cm <sup>3</sup> | Liquid<br>μ = 0.97 cP<br>d = 1.002 g/cm <sup>3</sup> | Liquid<br>μ = 0.54 cP<br>d = 0.990 g/cm <sup>3</sup>        | Liquid<br>μ = 0.45 cP<br>d = 0.986 g/cm <sup>3</sup>        |
| <b>CO<sub>2</sub> @1700psi</b> | Liquid<br>μ = 0.10 cP<br>d = 0.94 g/cm <sup>3</sup>  | Liquid<br>μ = 0.08 cP<br>d = 0.87 g/cm <sup>3</sup>  | Supercritical<br>μ = 0.042 cP<br>d = 0.56 g/cm <sup>3</sup> | Supercritical<br>μ = 0.029 cP<br>d = 0.38 g/cm <sup>3</sup> |



### CONCLUSIONS

Studying relative permeability properties of CO<sub>2</sub> and brine in reservoir rocks is of major importance for carbon sequestration.  
In the lab, we see that:  
-When the core is structured, the heterogeneities are controlling the spatial distribution of CO<sub>2</sub> at steady state. The less porous layers are hardly invaded.  
-When the sample is homogeneous and the injection flow rate is low enough, gravity effects become important and the CO<sub>2</sub> invades preferentially the top part of the core.  
-CO<sub>2</sub> saturation and relative permeability are seen to be flow rate dependent. At higher flow rate, CO<sub>2</sub> saturation is higher and relative permeability lower.  
- Numerical simulations are underway to validate these observations (see Chia-Wei Kuo's poster) as well as sub-core scale analysis (see Michael Krause's poster).